$\qquad$
Show your work, round to the correct significant figures, circle your answers, and label them with units.
When you have finished this, go to the website and check your answers. If you got a problem wrong, cross it off on the front, and do it correctly on the back.
2. What is the average kinetic energy of a molecule of an ideal gas at $20.0^{\circ} \mathrm{C}$ ? What is the total internal energy of 1.00 moles ( $6.02 \times 10^{23}$ molecules) of an ideal gas at this temperature?
3. At what temperature in Celsius is the average kinetic energy of an ideal gas molecule $5.00 \times 10^{-21} \mathrm{~J}$ ?
4. What is the RMS speed of a nitrogen molecule $(\mathrm{m}=2 \times 14=28.0 \mathrm{u})$ at $30.0^{\circ} \mathrm{C}$ ?
5. At what temperature in Celsius is the RMS speed of oxygen molecules $(\mathrm{m}=2 \mathrm{x} 16=32 \mathrm{u}) 470 . \mathrm{m} / \mathrm{s}$ ?

